

Page: 1 Printed: 26/03/2021 Revision: 26/03/2021

Supersedes Revision: 03/29/2016

#### 1. Product and Company Identification

Product Code: FGAU446-449
Product Name: BUDFEED PLUS
Trade Name: BUDFEED PLUS
Company Name: Stoller Australia Pty Ltd

1 Creswell Road

Largs Bay

South Australia 5016

Web site address: www.stoller.com.au stoller@stoller.com.au

Emergency Contact: STOLLER PRODUCTION CHEMIST

Contact number: 08 8169-0988

**Information:** 1800 337-845

Intended Use: For agricultural use only

#### 2. Hazards Identification

#### Serious Eye Damage/Eye Irritation, Category 2



GHS Signal Word: Warning

GHS Hazard Phrases: H319 - Causes serious eye irritation.

**GHS Precaution Phrases:** P264 - Wash hands thoroughly after handling.

P280 - Wear protective gloves/protective clothing/eye protection/face protection.

GHS Response Phrases: P305+351+338 - IF IN EYES: Rinse cautiously with water for several minutes. Remove

contact lenses, if present and easy to do. Continue rinsing. P337+313 - If eye irritation persists, get medical advice/attention.

GHS Storage and Disposal

Phrases:

No phrases apply.

Potential Health Effects Hazards not otherwise classified (HNOC) or not covered by GHS: None.

(Acute and Chronic): Chronic exposures to skin and mucous membranes that cause irritation may cause a

chronic dermatitis or mucosal membrane problem.

Inhalation: Inhaling mist, spray, or vapor may cause irritation to upper respiratory tract (nose and

throat). Nasal mucosal and oropharyngeal eythema.

**Skin Contact:** Skin irritation. Skin exposure may cause slight irritation, redness, itching, swelling. May

cause more severe response if skin is damp, abraded (scratched or cut), or covered by clothing, gloves or footwear. Prolonged contact may cause more severe symptoms.

Damage is localized to contact areas.

**Eye Contact:** Causes eye irritation. Eye exposure may cause serious eye irritation, pain, and/or

damage to the eye. May cause conjuntival swelling and cornea opacification from hypertonic solution. Corneal eye pain, redness, acute corneal thickening or whitening.

**Ingestion:** Consumption of hypertonic solutions causes nausea, vomiting, and increased thirst.

Medical Conditions Generally Any skin condition that disrupts the skin, such as abrasions, cuts, psoriasis, fungal

Aggravated By Exposure: infections, etc. Any eye condtion that compromises tear production, conjunctiva, or

normal corneal homestasis.

Page: 2 Printed: 26/03/2021 Revision: 26/03/2021

Supersedes Revision: 03/29/2016

3. Composition/Information on Ingredients				
CAS#	Components (Chemical Name)	Concentration		
22691-02-7	Calcium chloride (CaCl2), hydrate	<18.0 %		
7786-30-3	Magnesium chloride	< 5.0 %		
57-13-6	Urea	<40.0 %		

#### 4. First Aid Measures

**Emergency and First Aid** 

Procedures:

Victims of severe exposure to chemicals must be taken to health providing centers for medical attention. Always bring with victim a copy of label and SDS of product to health

professional.

In Case of Inhalation: Remove person to fresh air and keep comfortable for breathing. Call a POISON

CENTER or doctor/physician if you feel unwell.

Wash with plenty of water. If skin irritation occurs, get medical advice/attention. Take off In Case of Skin Contact:

contaminated clothing and wash before re-use.

In Case of Eye Contact: Rinse cautiously with water for several minutes. Remove contact lenses, if present and

easy to do. Continue rinsing. If irritation occurs, get medical advice/attention.

Rinse mouth. Call a POISON CENTER or doctor/physician if you feel unwell. In Case of Ingestion:

Signs and Symptoms Of

**Exposure:** 

Solution and/or solids may be visible on the skin and/or eyes. Localized redness, warmth, and irritation consistent with mechanism of injury: abrasion, burn, hypertonic

solution.

Treat symptomatically and supportively. Note to Physician:

#### 5. Fire Fighting Measures

N.A. Flash Pt:

LEL: N.A. UEL: N.A. **Explosive Limits:** 

N.A. **Autoignition Pt:** 

Suitable Extinguishing Media: Use extinguishing measures that are appropriate to local circumstances and the

surrounding environment.

Unsuitable Extinguishing

Media:

None known.

Fire Fighting Instructions: None specific for this product, however, it is suggested that firefighters wear

self-contained breathing apparatus (SCBA) and full protective equipment, such as

chemical resistant clothing.

Flammable Properties and

Hazards:

This material does not burn.

**Hazardous Combustion** 

Products:

Formed under fire conditions: hydrogen chloride gas, calcium oxide

#### 6. Accidental Release Measures

Protective Precautions, **Protective Equipment and Emergency Procedures:** 

Isolate the area. Keep unnecessary and unprotected personnel from entering the area. Spilled material may cause a slipping hazard on some surfaces. Use appropriate safety equipment. For addtional information, refer to Section 8, Exposure Controls and

Personal Protection. Refer to Section 7, Handling, for additional precautionary measures.

**Environmental Precautions:** Prevent entry into waterways, sewers, basements or confined areas. See Section 12,

Ecological Information.

Steps To Be Taken In Case Material Is Released Or

Spilled:

Small and large spills: Contain spilled material if possible. Absorb with materials such as sand. Collect in suitable and properly labeled containers. Flush residue with water. See

Section 13, Disposal Considerations, for additional information.

Page: 3 Printed: 26/03/2021 Revision: 26/03/2021

Supersedes Revision: 03/29/2016

## 7. Handling and Storage

Precautions To Be Taken in Handling:

Avoid contact with eyes, skin, and clothing. Do not swallow. Wash thoroughly after handling. Wear personal protective equipment as described in Section 8, Exposure

Controls/Personal Protection.

Precautions To Be Taken in Storing:

Protect from atmospheric moisture. Keep containers tightly closed when not in use. Keep separated from incompatible substances see Section 10, Stablility and Reactivity.

#### 8. Exposure Controls/Personal Protection

CAS#	Partial Chemical Name	OSHA TWA	ACGIH TWA	Other Limits
22691-02-7	Calcium chloride (CaCl2), hydrate	No data.	TLV: 10 mg/m³	No data.
7786-30-3	Magnesium chloride	No data.	No data.	No data.
57-13-6	Urea	No data.	No data.	No data.

Respiratory Equipment (Specify Type):

Respiratory protection should be worn when there is a potential to exceed the exposure limit requirements or guidelines. If there are no applicable exposure limit requirements or guidelines, wear respiratory protection when adverse effects, such as respiratory irritation or discomfort have been experienced, or where indicated by your risk assessment process. In dusty or misty atmospheres, use an approved particulate respirator. The following should be effective types of air-purifying respirators: high efficiency particulate air (HEPA) N95. A respiratory protection program that meets OSHA's 29 CFR 1910.134 and ANSI Z88.2 requirements or European Standard EN 149 must be followed whenever workplace conditions warrant respirator use.

**Eye Protection:** 

Wear safety glasses with side-shields. Wear chemical safety goggles and/or a

face-shield to protect against skin and eye contact when appropriate.

**Protective Gloves:** 

Use gloves chemically resistant to this material. If hands are cut or scratched, use gloves chemically resistant to this material even for brief exposures. Examples of preferred glove barrier materials include: Neoprene, Polyvinyl chloride ("PVC" or "vinyl"),

Notice: The selection of a specific glove for

Nitrile/butadiene rubber ("nitrile" or "NBR"). NOTICE: The selection of a specific glove for a particular application and duration of use in a workplace should also take into account all relevant workplace factors such as, but not limited to: other chemicals which may be handled, physical requirements (cut/puncture protection, dexterity, thermal protection), potential body reactions to glove materials, as well as the instructions specifications

provided by the glove supplier.

Other Protective Clothing:

Engineering Controls (Ventilation etc.):

Wear clean, body-covering clothing. Wear appropriate clothing to avoid skin contact.

Use local exhaust ventilation, or other engineering controls to maintain airborne levels below exposure limit requirements or guidelines. If there are no applicable exposure limit requirements or guidelines, general ventilation should be sufficient for most operations.

Local exhaust ventilation may be necessary for some operations.

Work/Hygienic/Maintenance Practices:

Use good personal hygiene. Do not consume or store food in the work area. Wash hands and affected skin immediately after handling, before smoking or eating, before breaks, and at the end of the workday.

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# **SAFETY DATA SHEET**

Page: 4

Printed: 26/03/2021 **BUDFEED PLUS** Revision: 26/03/2021 Supersedes Revision: 03/29/2016

	9. Physical and Chemical Properties		
Physical States:	[ ] Gas [ X ] Liquid [ ] Solid		
Appearance and Odor:	Light to medium dark amber color.		
	Very slight characteristic odor.		
pH:	6 - 8		
Freezing Point:	N.E.		
Boiling Point:	> 240.00 F		
Flash Pt:	N.A.		
Evaporation Rate:	N.E.		
Flammability (solid, gas):	Material will not burn.		
Explosive Limits:	LEL: N.A. UEL: N.A.		
Vapor Pressure (vs. Air or	N.E.		
mm Hg):			
Vapor Density (vs. Air = 1):	N.E.		
Specific Gravity (Water = 1):	1.2 - 1.3 at 20.0 C		
Density:	10.25 LB/GA at 20.0 C		
Solubility in Water:	Soluble		
Saturated Vapor	N.E.		
Concentration:			
Octanol/Water Partition Coefficient:	No data.		
Percent Volatile:	N.A.		
Autoignition Pt:	N.A.		
<b>Decomposition Temperature:</b>	N.A.		
Viscosity:	N.E.		
	10. Stability and Reactivity		
Stability:	Unstable [ ] Stable [ X ]		
Conditions To Avoid - Instability:	Stable under normal temperatures and pressures.		
Avoid:	Avoid contact with: bromide trifluoride. 2-furan percarboxylic acid because calcium chloride is incompatible with those substances. Contact with zinc forms flammable hydrogen gas, which can be explosive. Catalizes exothermic polymerization of methyl vinyl ether. Reaction of bromide impurity with oxidizing materials may generate trace levels of impurities such as bromates.		
Hazardous Decomposition or Byproducts:	Formed under fire conditions: hydrogen chloride gas, calcium oxide		
Possibility of Hazardous Reactions:	Will occur [ ] Will not occur [ X ]		
Conditions To Avoid - Hazardous Reactions:	None known.		

Printed: 26/03/2021 Revision: 26/03/2021 Supersedes Revision: 03/29/2016

#### 11. Toxicological Information

**Toxicological Information:** Mutagenicity: This product has not been investigated for mutagenic effects.

Embryotoxicity: This product has not been investigated for embryotoxic effects.

Teratogenicity: This product has not been investigated for teratogenic effects.

Reproductive Toxicity: This product has not been investigated for toxic reproductive

effects.

CAS# 57-13-6: Acute toxicity, LD50, Oral, Rat, 8471. MG/KG. Result: Autonomic Nervous System: Other (direct) parasympathomimetic. Behavioral: Coma.

Gastrointestinal: Hypermotility, diarrhea.; Gigiena i Sanitariya, Mezhdunarodnaya Kniga,

ul. B. Yakimanka, 39, 113095, Moscow 113095 Russia, Vol/p/yr: 51(6),8, 1986

Irritation or Corrosion: Symptoms related to

No data available. No data available.

Toxicological Characteristics:

NO data available

Sensitization:

No data available.

Chronic Toxicological

The toxicological properties of this material have not been fully investigated.

Effects:

No component is listed as a carcinogenic by IARC, NTP, OSHA, and ACGIH.

Carcinogenicity/Other Information:

Carcinogenicity:

NTP? No IARC Monographs? No OSHA Regulated? No

# 12. Ecological Information

General Ecological Information:

The available data on this material does not indicate any undue hazard to the environment under anticipated use and storage. All work practices must be aimed at eliminating environmental contamination. Any waste due to spillage or leakage should be

contained and disposed of accordingly, see above under Section 6 "Accidental Release

Measures."

Results of PBT and vPvB

assessment:

No data available.

Persistence and

Degradability:

Calcium chloride is believed not to persist in the environment because it is readily dissociated into calcium and chloride ions in water. Both ions originally exist in nature, and their concentrations in surface water will depend on various factors, such as

geological parameters, weathering and human activities.

Bioaccumulative Potential: Calcium chloride and its dissociated forms (calcium and chloride ions) are ubiquitous in

the environment. Calcium and chloride ions can also be found as constituents in organisms. Considering its dissociation properties, calcium chloride is not expected to

accumulate in living organisms.

**Mobility in Soil:** Chloride ions are mobile in soil eventually drainig into surface water.

#### 13. Disposal Considerations

Waste Disposal Method: PRODUCT: Reuse or reprocess, if possible. Waste disposal must be done following all

Federal, State and Local regulations. Regulations may vary in different locations. Report spills if applicable. Chemical waste generators must determine whether a discarded chemical is classified as a hazardous waste. This product, if unaltered by use, may be disposed of by treatment at a permitted facility or as advised by your local waste

regulatory authority.

CONTAINER: Dispose properly accordingly to regulations on empty containers in your

locality or make available to a container reconditioning facility for recycling.

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Page: 5

Page: 6 Printed: 26/03/2021 Revision: 26/03/2021

Supersedes Revision: 03/29/2016

# 14. Transport Information

LAND TRANSPORT (US DOT):

**DOT Proper Shipping Name:** Not Regulated.

Trade Name: BUDFEED PLUS

DOT Hazard Class: UN/NA Number:

MARINE TRANSPORT (IMDG/IMO):

IMDG/IMO Shipping Name: Not Regulated. Trade Name: BUDFEED PLUS UN Number: Packing Group:

**Hazard Class:** 

IMDG MFAG Number: N.A.

**IMDG EMS Page:** 

AIR TRANSPORT (ICAO/IATA):

ICAO/IATA Shipping Name: Not Regulated. Trade Name: BUDFEED PLUS

Additional Transport

Placards / Markings: N.A.

Information:

Emergency Response Guide Number: N.A.

Reportable Quantity: N.A.

# 15. Regulatory Information

# 16. Other Information

**Revision Date:** 26/03/2021

HEALTH 1
FLAMMABILITY 0
REACTIVITY 0
PPE

HMIS:

Flammability Instability
Health
NFPA: Special Hazard

Additional Information About No data available.

This Product:

Company Policy or

**Hazard Rating System:** 

Disclaimer: